Contamination of Polycyclic Aromatic Hydrocarbons Bound to PM₁₀/PM_{2.5} in Xiamen, China

X.H. Wang^{1*}, C.X. Ye¹, H.L. Yin¹, M.Z. Zhuang², S.P. Wu¹, J.L. Mu¹, H.S. Hong¹

¹ State Key Laboratory of Marine Environmental Science/Environmental Science Research Center, Xiamen University, Xiamen, China ² Xiamen Environmental Monitoring Central Station, Xiamen, China

Abstract

 PM_{10} (particulate matter which passes through a size selective impactor inlet with a 50% efficiency cut-off at 10 µm aerodynamic diameter) and $PM_{2.5}$ were collected at four sites (A: Xiaoping, B: Huli, C: Hongwen, D: Gulang Island) from December 20, 2004 to December 29, 2004 in Xiamen. 16 Priority polycyclic aromatic hydrocarbons (PAHs) in $PM_{10}/PM_{2.5}$ were quantified by using a gas chromatograph/mass spectrometer (GC/MS). EC and OC of PM_{10} were analyzed by Thermal/Optical Carbon Analyzer (DRI Model 2001). Concentration, spatial distributions of PAHs and EC/OC were studied and their main sources were identified by diagnostic ratios, correlation analysis and EC/OC analysis. Among the different sampling sites, concentration of OC were 9.83-15.6 µg/m³ and 1.41-4.73 µg/m³ for EC associated to PM_{10} . The main sources identified were traffic emissions and industry emission/coal combustion.

Keywords: PAHs; EC/OC; Source identification; Xiamen.

Corresponding author. Tel.: 86-592-2187857, Fax: 86-592-2180655

E-mail address: xhwang@xmu.edu.cn

INTRODUCTION

Polycyclic aromatic hydrocarbons (PAHs) are ubiquitous pollutants in the air which are well-known to be toxic, carcinogenic and mutagenic (Ames *et al.*, 1975; MØller *et al.*, 1982). PAHs bound to PM_{10} are considered to be a great threat to the health of human beings because they could be brought into the body through breathing (Spurny, 1996). Many studies indicate that fine particles ($PM_{2.5}$) have greater impact on human health than coarse particles (Schwartz, 1994). Numerous studies were focused on the concentrations and sources of PAHs bound to $PM_{10}/PM_{2.5}$ in the urban air (Gao *et al.*, 1993; Qi *et al.*, 2000; Guo *et al.*, 2003; Peng *et al.*, 2003; Pan *et al.*, 2003; Guo *et al.*, 2004; Dallarosa *et al.*, 2005; Ding *et al.*, 2007) and found it was important to study the abundance, distribution and potential sources of PAHs associated to $PM_{10}/PM_{2.5}$ so that the diverse effects caused by particulate matter could be efficiently controlled.

Xiamen is located along the southeastern shore of China at E117°53'-118°27', N24°25'-24°55', as shown in Fig.1. It has a population of about 1.4 million people living in a terrene-area of 1565.09 km² and a sea-area of 344 km². It includes Xiamen Island, Gulang Island, Tongan, Jimei, Haicang, Xinglin and many other small islands, amongst which Xiamen Island has an area of 132.5 km² and a 234 km coastline. Xiamen has been a famous "garden city" due to its clear sky and beautiful environment, but nowadays its air quality decreased from "excellent" to "good", with the main pollutants being PM₁₀. Other studies (Karar and Gupta, 2006; Valavanidis *et al.*, 2006) showed that the highest concentrations of PAHs in particulate phase always occurred in winter. In this study, we aimed to carry out the wintertime study of concentrations, distribution patterns and main sources of PAHs bound to PM₁₀/PM_{2.5} in Xiamen.



Fig. 1. Map of Xiamen with sampling sites shown as large dots.

METHODS

Sample collection

Sampling of PM₁₀ and PM_{2.5} were carried out at several sites (A, B, C, D) during December 21-29, 2004 in Xiamen, Fujian province, P.R. China (Fig. 1). The sampling sites were selected to present the different functional areas: Site A: Xiaoping (N 24°51.218', E118°2.689', 900 m elevation), acted as the control site located in a forest garden to the north of Xiamen island. There is no major pollution source around 25 km area except diesel engine vehicles were used to build the road. Site B: Huli (N24°30.51', E118°6.139', 8 m elevation), as industry/traffic area. The sampling site was located on the roof of an official building, about 100 m away from a road carrying heavy traffic (1300/h, 2800/h) and surrounded by a series of industries, such as tobacco production plant, aluminium plant, building material production, etc. Site C: Hongwen (N24°28.623, E118°8.703, 15 m elevation). The sampling site was located on the roof of a residential area, with no other high buildings or industry nearby. Site D: Gulang Island (N24°26.787, E118°3.819, 10 m elevation). The island is separated from Xiamen Island with a 5 min boat ride which is famous for tourism. There is no industry or traffic except for battery driven vehicles on the island. The sampling site was on the roof of a school building, about 500 m from the entrance to the island.

Daily sampling at the four sites began almost simultaneously at 10:00 a.m., with a 23 h duration. PM_{10} samplers (100 L/min, Qingdao Laoshan Mechanical Corp., China) with Ø 90 mm glass fiber filters (GFFs, Whatman Corp. USA) and $PM_{2.5}$ sampler (MiliVol Airmetrics, USA, 5 L/min) with Ø 47 mm glass fiber filters (GFFs, Whatman Corp. USA) were used to collect aerosols. The samples deposited on the filters were analyzed for 16 priority PAHs, and EC/OC for PM_{10} . Due to the limitation of the available $PM_{2.5}$ sampler, only two $PM_{2.5}$ samples at each site were gotten. So the following discussion was more focused on PM_{10} .

Sample analysis

The glass fiber filters were pre-baked at 450 °C for 4 h to lower their carbon blanks. GFF samples were spiked with deuterated PAHs standards (naphthalene-d₈, acenaphthene-d₁₀, acenaphthene-d₁₀, chrysene-d₁₂, perylene-d₁₂) prior to sample extraction. The filter was extracted with 20 mL aliquots of dichloromethane in the dark overnight, and then ultrasonically extracted for 30min at room temperature, twice. The total extract was concentrated to a volume of 1 mL by vacuum-condensing equipment. The cleanup and fractionation of the extract was done using a self-packed silica gel-Al₂O₃-Na₂SO₄ column, a 5 mL acetone-rinsed burette containing 3 mL firmly packed deactivated silica gel (72-230 mesh, Sigma) in the bottom, a layer of 2 mL firmly packed deactivated Al₂O₃, and then 1 cm Na₂SO₄ on the top. Ten milliliters hexane was added to condition the column. The concentrated extract was then applied to the column with an additional

3 mL hexane for complete transfer. n-Hexane was used to elute the n-alkanes fraction (5 mL, F1), benzene: n-hexane (v/v, 1:1) was used to elute the aromatic fraction (10 mL, F2). The volume of each fraction was concentrated to near dryness by vacuum-condensing equipment, and then fixed to 100 μ L. The PAHs in F2 fractions were analyzed by GC-MSD (HP 6890 GC–HP 5973 MSD) in Selected Ions Mode (SIM). The GC was equipped with a HP-5 MS capillary column (30 m × 0.25 mm i.d., film thickness 0.25 μ m), with helium as carrier gas. The oven temperature program was as follows: hold at 60°C for 2 min, increased to 120°C at a rate of 4°C/min, then increased to 300°C at a rate of 3°C/min and a final hold at 300°C for 10 min. The injector and detector temperature were 250°C and 280°C. 1 μ L sample was injected in a splitless model. The mass scanning ranged between 50 and 500 m/z.

16 PAHs were quantified in SIM mode according to their elution orders as followed: naphthalene (Nap, m/z = 128), acenaphthylene (Ace, m/z = 152), acenaphthene (Acen, m/z = 154), fluorene (Flu, m/z = 166), phenanthrene (Phen, m/z = 178), anthracene (An, m/z = 178), fluoranthene (Fluo, m/z = 202), pyrene (Pyr, m/z = 202), benzo(a)anthracene (BaA, m/z = 228), chrysene (Chry, m/z = 228), benzo(b)fluoranthene (BbF, m/z = 252), benzo(k)fluoranthene (BkF, m/z = 252), benzo(a)pyrene (BaP, m/z = 252), indeno(1,2,3,-cd)pyrene (Indeo, m/z = 276), dibenzo(a,h)anthracene (DiBa, m/z = 278), benzo(ghi)perylene (BgP, m/z = 276).

EC/OC were analyzed by National Environmental Analysis Center using Thermal/Optical Carbon Analyzer (DRI Model 2001) with the method detection limits (MDL) of 0.82 μ g C/cm² for OC and 0.19 μ g C/cm² for EC.

Quality control

The analytical method was based on the USEPA Method TO-13 (US EPA, 1999). Field blanks, filter blanks and solvent blanks were analyzed by the same procedure as the samples and ensured that there were no significant background interferences with only trace amount of PAHs being detected out (ND-5.13 ng). When compared with the amount (160.94-748.88 ng) in the real samples, the concentration in blank samples could be neglected.

The quantifications were performed using the internal standard method. The recovery test was performed by spiking known amounts of a mixture of PAHs onto the GFFs and then the spiked filter was treated the same way as mentioned above. Method detection limit (MDL) was calculated taking into account 3 times the ratio of the heights obtained for noise. Recovery efficiency of PAHs ranged from 60.1% to 119.8%. Method detection limits were between 1.5×10^{-3} ng/m³ (BkF) and 11.4×10^{-3} ng/m³ (Nap), as shown in Table 1.

	5 ()	
Compounds	RR (%)	MDL (× 10^{-3} ng/m ³)
Nap	109.4	11.40
Ace	83.5	5.86
Acen	87.7	6.61
Flu	83.6	5.08
Phen	111.3	2.56
An	93.4	3.08
Fluo	60.1	1.75
Pyr	61.0	1.65
BaA	111.1	2.03
Chry	108.5	1.60
BbF	94.3	1.81
BkF	98.7	1.51
BaP	110.2	2.46
Indeo	118.8	2.83
DiBa	119.8	3.00
BgP	105.5	2.02

Table 1. The PAHs recovery ratio (RR) and method detection limits (MDL).

RESULTS AND DISCUSSION

Comment on meteorology and PAHs concentrations

The meteorological conditions and air pollution index during the sampling periods were shown in Table 2. Temperature was about 8-20°C, wind speed was almost higher than 2 m/s and relative humility was high (71%-92%) which were the typical subtropical climate. The relationship between different PAHs bound to PM₁₀ and meteorology parameters were analyzed by SPSS packages (Table 3). Σ PAHs have inverse correlation with relative humidity (-0.538) and weak correlations with temperature (r = 0.34) or wind speed (r = 0.42) (p < 0.05). But PM₁₀ has significant correlation with temperature with r = 0.74 (p < 0.05) while inverse correlation with wind speed (r = -0.005) and RH (r = -0.20). Thus is consistent with results observed by Karar and Gupta (2006) in Kolkata, India and Mantis *et al.* (2005) in the Greater area of Athens, Greece. When relative humidity increases, the fine particulates have more chances to aggregate together and grow bigger, the concentration of PM₁₀ decreased and also of Σ PAHs bound to PM₁₀.

	Tuble 2. Meteorological conditions during the sampling duration.						
Date	Temp. range	Average Temp.	Wind direction	Wind speed	Relative humidity		
	(°C)	(°C)	while direction	(m/s)	(%)		
2004-12-20	12-21	16.0	NE35°	3.0	79.2		
2004-12-21	14-22	16.6	NNE18°	3.4	77.0		
2004-12-22	13-23	17.1	E100°	1.5	84.2		
2004-12-23	13-22	16.0	ESE107°	2.0	91.7		
2004-12-24	11-20	14.7	ESE103°	2.2	83.2		
2004-12-25	12-19	15.0	ESE106°	2.5	81.5		
2004-12-26	12-19	15.3	ESE105°	2.2	85.7		
2004-12-27	13-19	14.9	WNW288°	2.4	90.5		
2004-12-28	12-17	11.6	NE35°	2.1	88.0		
2004-12-29	8-14	9.7	NNE18°	4.5	71.0		

Table 2. Meteorological conditions during the sampling duration.

Table 3. Correlation coefficient of meteorological conditions and Σ PAHs in PM₁₀.

	∑PAHs	PM_{10}	Temp	Wind	RH
∑PAHs	1				
PM_{10}	0.810^{**}	1			
Temp	0.337	0.735^{*}	1		
Wind	0.422	-0.005	-0.598	1	
RH	-0.538	-0.204	0.404	-0.826**	1

* Correlation is significant at the 0.05 level (2-tailed); ** Correlation is significant at the 0.01 level (2-tailed).

Conc. = concentration, Temp. = temperature, RH = relative humidity.

In Xiamen, 16 PAHs were detected out in all samples except that Ace was below the detection limit (Table 4). In PM₁₀, the concentration range of individual PAHs was from ND-13.31 ng/m³ and 5.20-28.07 ng/m³ for Σ 16PAHs. The highest concentration of Σ 16PAHs almost occurred in the first sampling day (Dec. 20) at different sites. From the second sampling day, the variation of total concentrations was not much suggesting the stable pollution level under similar meteorology conditions. Among PAHs, Phen (2.16-13.31 ng/m³), Flu (0.21-7.04 ng/m³), Pyr (0.53-2.44 ng/m³) and Chry (0.40-1.75 ng/m³) were main species. They are all lower and medium molecular weight (MW \leq 228) PAHs. In PM_{2.5}, the concentration range of Σ 16PAHs was 3.04-11.32 ng/m³. Among PAHs, the main species not only include low and middle molecular weight PAHs such as Phen (0.92-1.78 ng/m³), Pyr (0.38-1.97 ng/m³) and Chry (0.51-1.60 ng/m³), etc., but also some higher molecular weight (MW \geq 252) PAHs such as BaP (0.16-1.34 ng/m³), Indeo (0.18-1.30 ng/m^3) and BgP (0.22-1.52 ng/m^3). These results suggest the higher molecular weight PAHs were more easily bound to fine particulates.

Table 4. Concentrations and ratios of PAHs in PM_{10} and $PM_{2.5}$ at different sampling sites (ng/m³).

PM ₁₀				PM _{2.5}				
Conc.	А	В	С	D	А	В	С	D
Nap	0.01 ± 0.02	0.01 ± 0.02	0.01 ± 0.01	0.01 ± 0.01	0.01	0.03	0.06	0.05
Ace	ND	ND	ND	ND	ND	ND	ND	ND
Acen	0.07 ± 0.02	0.23 ± 0.13	0.10 ± 0.04	0.11 ± 0.07	ND	ND	ND	ND
Flu	0.39 ± 0.16	1.53 ± 2.07	0.56 ± 0.30	1.17 ± 1.91	0.12	0.15	0.14	0.14
Phen	3.40 ± 1.63	8.24 ± 2.50	3.31 ± 1.61	5.07 ± 1.85	1.16	1.39	0.92	1.78
An	0.20 ± 0.11	0.22 ± 0.12	0.22 ± 0.12	0.28 ± 0.12	ND	0.11	0.11	0.11
Fluo	0.31 ± 0.06	0.63 ± 0.21	0.31 ± 0.12	0.54 ± 0.14	0.13	0.4	0.22	0.45
Pyr	0.79 ± 0.17	1.75 ± 0.46	0.99 ± 0.39	1.49 ± 0.41	0.38	1.97	0.96	1.45
BaA	0.10 ± 0.06	0.34 ± 0.12	0.29 ± 0.13	0.26 ± 0.17	0.58	1.6	0.89	0.51
Chry	0.54 ± 0.13	0.81 ± 0.43	0.64 ± 0.17	0.57 ± 0.15	ND	ND	ND	ND
BbF	0.13 ± 0.03	0.20 ± 0.09	0.25 ± 0.06	0.22 ± 0.10	0.1	0.65	0.25	0.15
BkF	0.14 ± 0.05	0.25 ± 0.10	0.29 ± 0.08	0.26 ± 0.17	ND	0.81	0.31	0.17
BaP	0.24 ± 0.07	0.51 ± 0.24	0.48 ± 0.18	0.45 ± 0.32	0.16	1.34	0.46	0.22
Indeo	0.28 ± 0.06	0.51 ± 0.19	0.50 ± 0.13	0.45 ± 0.23	0.18	1.3	0.57	0.26
DiBa	0.01 ± 0.00	0.02 ± 0.01	0.02 ± 0.01	0.02 ± 0.01	ND	0.05	0.03	ND
BgP	0.26 ± 0.06	0.45 ± 0.19	0.46 ± 0.11	0.41 ± 0.17	0.22	1.52	0.62	0.3
ΣPAHs	6.81 ± 1.97	16.09 ± 5.10	8.34 ± 2.42	11.20 ± 4.11	3.04	11.3	5.54	5.59
Fluo/Pyr	0.39 ± 0.02	0.36 ± 0.03	0.32 ± 0.05	0.36 ± 0.04	0.35	0.2	0.23	0.31
BaA/Chry	0.18 ± 0.08	0.50 ± 0.22	0.44 ± 0.14	0.43 ± 0.18	-	-	-	-
Pyr/BaP	3.51 ± 1.12	4.17 ± 2.35	2.16 ± 0.74	4.17 ± 1.80	2.3	1.48	2.11	6.5

ND = not detected. SD = standard deviation.

Compared with other cities in China, concentration of Σ PAHs associated to PM₁₀ in Xiamen was significantly lower than those in inland cities, such as Beijing (116.0 ng/m³ for annual average concentration) (Zhou *et al.*, 2005) and Shanghai (245.52 ng/m³ for winter) (Guo *et al.*, 2004), but in the similar level as the southern coastal cities including Guangzhou (55.49 ng/m³ for winter and spring), Hongkong (14.88 ng/m³ for winter and spring) and Macao (41.70 ng/m³ for winter and spring) (Qi *et al.*, 2000), etc. One of the most obvious reasons for this is that central or residential heating in winter is a major pollution source in the northern inland cities (Beijing, etc.) while southern cities do not have this pollution source. Compared with overseas

cities, concentration of Σ PAHs in Xiamen was lower than that in New Delhi(Sharma *et al.*, 2003), but in the same level of that in some coastal cities, such as Naples (Caricchia *et al.*, 1999), Athens (Mantis *et al.*, 2005) and Santiago (Romero, 2002), etc.

Concentration of BaP is an indicator for the control of Σ PAHs in the atmosphere. Its acceptable concentration limit is 10 ng/m³ based on the Chinese standard (GB3095-1996). The concentrations of BaP in PM₁₀ and PM_{2.5} were in the range 0.10-1.34 ng/m³ meaning that the PAHs pollution was not serious in Xiamen.

For better quantification of whole carcinogenicity from whole PAH fraction, "BEQ" (Benzo[a]pyrene Equivalence) was introduced as an index and calculated by the following formula using the toxic equivalency factors (TEFs) of the individual PAHs (Robert *et al.*, 1999) and their concentrations:

$$BEQ = BaP \times 1 + BaA \times 0.12 + An \times 0.002 + Phen \times 0.028 + Fluo \times 0.033 + BbF \times 0.15 + Pyr \times 0.046 + Chry \times 0.25.$$
 (1)

The BEQ index was 0.85 and 0.92 for PAHs in PM_{10} and $PM_{2.5}$ in Xiamen which was much lower than that of Shanghai (15.77) (Guo *et al.*, 2004), but in the same level of that in some Japanese cities (1.91) (Takeshi *et al.*, 2004).

Relationship of concentration of Σ PAHs and concentration of PM₁₀ was analyzed by SPSS (Table 3). Results showed that there were significant correlation (r = 0.71, p < 0.01, n = 35) between the concentration of Σ PAHs and PM₁₀. These results suggested that sources of PM₁₀ and PAHs in the air were similar, and the control of PM₁₀.was an effective way to decrease the PAHs pollution level in the air as well.

Distribution of PAHs bound to PM₁₀/PM_{2.5}

Among different function areas, there were some difference existed in the concentration and distribution patterns of PAHs in $PM_{10}/PM_{2.5}$ (Fig. 2-Fig. 5, Table 4). The concentration ranges and standard deviation of Σ 16PAHs bound to PM_{10} at different sampling sites were as follows: B: 11.0-28.1 ng/m³ (16.1 ± 5.1 ng/m³) > D: 7.8-21.1 ng/m³ (11.2 ± 4.1 ng/m³) > C: 6.6-13.5 ng/m³ (8.4 ± 2.5 ng/m³) > A: 5.2-11.5 ng/m³ (6.9 ± 2.0 ng/m³), respectively. The concentration of Σ 16PAHs bound to PM_{2.5} showed the same patterns: B (11.32 ng/m³) > D (5.59 ng/m³) > C (5.54 ng/m³) > A (3.04 ng/m³).



Fig. 2. Concentrations of $\Sigma 16PAHs$ in PM_{10} at four sites of Xiamen.



Fig. 3. Average concentration of Σ 16PAHs bound to PM₁₀/PM_{2.5} at four sites of Xiamen.



Fig. 4. Distribution of PAHs in PM₁₀ at different sites.



Fig. 5. Distribution of PAHs in PM_{2.5} at different sites.

In PM₁₀ and PM_{2.5}, the highest concentration of Σ 16PAHs always occurred at site B (industry/traffic area) which was quite related to its environmental conditions. Site B was 100 m away from the main line carrying heavy traffic (1300/h, 2800/h) and also surrounded by a series of industries, such as tobacco production plant, aluminium smelting plant, etc. Large amount of industry and traffic emissions made the concentration of PAHs highest at this site. Among PAHs, Flu and Phen covered much more of total PAHs at this site than at other sites in PM₁₀. This was consistent with the fact that this area was a dense traffic area with more contribution of heavy-duty vehicles emissions (Marr *et al.*, 1999; Rogge *et al.*, 1993). In PM_{2.5}, BaP, Indo and BgP covered more of Σ 16PAHs at site B than at other sites meaning that there is more

contribution from gasoline vehicles (Simicik *et al.* , 1999) at site B. The concentration of PAHs at site D (Gulang Island) was ranked second. Though there was no direct automobile emission or industry emission, the heavy ship traffic to this famous tourist island and the wind bringing with the particulates from the southwest SongYu coal-generating electricity plant and the influence of the heavy road traffic emissions from the thither area could not be neglected. Site A (forest garden area) and site C (residential area) were relatively clean which were in consistent with the fact that there were no heavy traffic or industry nearby. Also plants could effectively clean out PAHs in the atmosphere (Simonich *et al.*, 1994).

Though the concentration of PAHs was quite different, the distribution patterns of PAHs bound to PM₁₀ in different function areas were quite similar. At all sites, 3-4 ring PAHs including Phen (39.2%-51.2%), Flu (5.7%-10.4%), Pyr (10.9%-11.8%) and Chry (5.0%-7.9%) were the dominant PAHs components bound to PM₁₀ (Fig. 4). These four PAHs covered 60.8%-81.3% of Σ 16PAHs. These distribution profiles were quite different from many other cities (Papageorgopoulou *et al.*, 1999; Caricchia *et al.*, 1999; Chetwittachan *et al.*, 2002; Pan *et al.*, 2003; Manoli *et al.*, 2004; Christine *et al.*, 2005), but similar with that in Athens, Greece which pointed out the great contribution from traffic emissions (Valavanidis, 2006). In PM_{2.5}, some higher molecular weight PAHs such as BaP (5.3-11.8%), Indeo (4.7-11.5%), BgP (5.4-13.4) covered more than that of PM₁₀. Among low and middle molecular PAHs, Phen was the dominant component which covered 12.3-38.2% of Σ 16PAHs, Pyr, Chry and Fluo covered 12.5-25.9%, 9.1-19.1% and 3.5-8.1% of Σ 16PAHs, respectively.

PM_{10}				PM _{2.5}				
r	А	В	С	D	А	В	С	D
А	1.000				1.000			
В	0.994 **	1.000			0.514	1.000		
С	0.993 **	0.986 **	1.000		0.756^{*}	0.945**	1.000	
D	0.990 **	0.995 **	0.990 **	1.000	0.843**	0.597^{*}	0.780^{*}	1.000

Table 5. Correlations of standardized concentration of PAHs at different sampling sites.

* Correlation is significant at the 0.05 level (2-tailed); ** Correlation is significant at the 0.01 level (2-tailed).

SPSS packages were used to detect the correlation of distribution patterns of individual PAHs at different sampling sites. Results showed that there was no significant difference in the distribution of PAHs in different function areas (Table 5). In PM₁₀, all the values of correlation coefficient (r) were much higher than 0.9 (P = 0.01) meaning that the sources of PAHs were quite similar at different sampling sites. In PM_{2.5}, all the values of r were higher than 0.50, in which values of 0.95, 0.84 and 0.76 were for PAHs at site B and C, A and D, A and C, respectively.

Thus could be explained by the fact that Xiamen was a coastal city with relative simple energy structure and the wind from the sea could better dilute and diffuse the pollutants in the air.

Source identification of PAHs in PM₁₀/PM_{2.5}

Diagnostic ratio

PAHs have been used as tracers to distinguish between diverse sources (Venkataraman and Friedlander, 1994; Mandalakis *et al.*, 2002; Guo *et al.*, 2005). Different profiles and distribution patterns of individual PAHs were characteristic of different pollution sources. As a result of numerous studies, Kulkarni and Venkataraman (2000) identified some PAHs as tracers of several sources in the urban atmosphere: Phen, Fluo and Pyr are characteristic of coal combustion; Pyr, Phen, Fluo are derived from incineration processes; BaP and Fluo are tracers for wood combustion; BgP and Indeo for vehicular fuels; and Fluo, Pyr, BbF and BkF for heavy-duty diesel vehicles. The main PAHs sources in the Asian area were traffic exhausts (Acen, Fluo, Flu, Pyr, Chry, BeP) and industrial emissions (BaP, BaA, BeP, Cor) (Chang *et al.*, 2005).

The sum of the concentrations of the nine following combustion PAHs (CPAHs): Fluo, Pyr, BaA, BbF, BkF, BaP, BeP, Indeo and BgP has often been used as a characteristic diagnostic parameter for PAHs produced by stationary combustion sources (Rogge, 1993; Kavouras, 1999; Bi et al., 2002; Manoli et al., 2002; 2004; Sienra, 2005), Rogge et al. (1993) determined the value of CPAHs/2PAHs ratio for non-catalyst (0.4) and catalyst-equipped (0.5) automobiles and for heavy-duty diesel trucks (0.3). In this study, the value of CPAHs/ Σ PAHs in PM₁₀ was 0.3-0.4 meaning that stationary combustion sources (industry emission) was a contributor for the PAHs pollution in the air. In PM_{2.5}, its value was 0.4-0.7 meaning that stationary combustion sources contribute much more than that at PM₁₀. Lee et al. (1977) suggested value of Fluo/Pyr for coal combustion (1.4) and for wood burning (1.0) and for fuel combustion (<1). Gschwend *et al.* (1981) indicated value of BaA/Chry ratio was 0.93 for wood burning and 0.28-1.2 for gasoline combustion and 0.17-0.36 for coal combustion. In this study, the values of Fluo/Pyr both in PM_{10} and PM_{2.5} were 0.20-0.39 indicating the fuel combustion source especially traffic emissions source was a great contributor; BaA/Chry in PM₁₀ was 0.18-0.50 meaning the gasoline and diesel vehicles emission sources. Thus most of the diagnostic ratios in PM₁₀ and PM_{2.5} suggested the significant contribution of gasoline exhaust and diesel exhaust, with some contribution from coal and wood combustion sources.

OC/EC and inorganic analysis

Carbonaceous particles consist of two major components, elemental carbon (EC) and organic carbon (OC), the sum of them being defined as total carbon (TC). Elemental carbon (EC) was associated with primary emission from fossil fuel combustion and wood burning. Because of its stability in the atmosphere, EC was usually used as the marker for anthropogenic pollution.

Organic carbon (OC) can be emitted directly from the emission sources or produced from atmospheric transformation of organic gases. Secondary OC is formed by condensation of low vapor pressure products during photo-oxidation of hydrocarbons (Seinfeld *et al.*, 1998; Chow *et al.*, 1996).

In this study, OC/EC in PM₁₀ was analyzed. Concentrations of OC were in the range of 9.83-15.60 μ g/m³ and 1.41-4.73 μ g/m³ for EC in Xiamen (Wang *et al.*, 2006). Among different function areas, the concentrations of OC were as following: B (15.60 μ g/m³) > D (11.83 μ g/m³) > C (10.99 μ g/m³) > A (9.83 μ g/m³) and the concentrations of EC were: B (4.73 μ g/m³) > C (3.13 μ g/m³) > D (2.86 μ g/m³) > A (1.41 μ g/m³). Ratios of OC/EC were as following: 6.98 (A) > D (4.13) > C (3.52) > B (3.30). These results meant the secondary organic carbon (SOC) contributed much to the PM₁₀ of Xiamen with all ratios higher than 3. Its contribution was calculated by the following equation:

$$SOC = TC - EC \times (OC/EC)_{min.}$$
 (2)

Results showed the secondary OC covered 38.1% in average of the total OC in Xiamen atmosphere. Generally, these results could be explained by the mild meteorology conditions with high temperature and good sunshine even in winter. They were quite suitable for the photochemical reactions to generate the SOC. The differences among sites were quite consistent with the fact that the pollutants at site B were more contributed by the continuous fresh input from traffic emission and industry emission while site A and D were more contributed by the secondary pollution due to the lack of the primary emission sources.

Concentration of Σ PAHs and OC significantly correlated with r = 0.59 (p < 0.01) while correlation coefficient between EC and OC was 0.80 (p < 0.01). Thus meant EC and OC were from similar sources (Offenberg *et al.*, 2000). In conjunction with the results of the high accumulation level of S, Se, Zn and Br in PM₁₀.by the elemental and inorganic analysis (Zhuang *et al.*, 2006), the mixing pollution sources of coal combustion/industry emission and serious traffic emissions were identified.

CONCLUSIONS

Wintertime study of atmospheric concentration of Σ 16PAHs bound to PM₁₀/PM_{2.5}, EC/OC were carried out during Dec. 21, 2004-Dec. 29, 2004. Concentrations of Σ 16PAHs were 5.20-28.07 ng/m³ in PM₁₀ and 3.04-11.32 ng/m³ in PM_{2.5}. Concentration of OC was 9.83-15.6 µg/m³ and 1.41-4.73 µg/m³ for EC. The highest concentration of Σ 16PAHs always occurred at site B meant the industry emission and traffic emission were great contributor for PAHs pollution

in Xiamen. Ratio analysis, EC/OC analysis and correlation analysis suggested the main sources of PAHs in $PM_{10}/PM_{2.5}$ were traffic emissions, industry emission and coal combustion.

REFERENCES

- Ames, B.N., Mccann, J. and Yamasaki, E. (1975). Method for Detecting Carcinogens and Mutagens with the Salmonella/Mammalian - Microsome Mutagenicity Test. *Mutat. Res.* 31: 347-364.
- Bi, X.H., Sheng, G.Y. and Peng, A.P. (2002). Extractable Organic Matter in PM₁₀ from LiWan District of Guangzhou City, P.R. China. *Sci. Total Environ.* 300: 13-228.
- Caricchia, A.M., Chiavarini, S. and Pezza, M. (1999). Polycyclic Aromatic Hydrocarbons in the Urban Atmospheric Particulate Matter in the City of Naples (Italy). *Atmos. Environ.* 33: 3731-3738.
- Chang, K.F., Fang, G.C., Chen, J.C. and Wu, Y.S. (2005). Atmospheric Polycyclic Aromatic Hydrocarbons (PAHs) in Asia: A Review from 1999 to 2004. *Environ. Pollut.* 142: 388–396.
- Chetwittachan, T., Shimazaki, D. and Yamamoto, K. (2002). A Comparison of Temporal Variation of Particle Bound Polycyclic Aromatic Hydrocarbons Concentration in Different Urban Environments: Tokyo, Japan and Bangkok, Thailand. *Atmos. Environ.* 36: 2027-2037.
- Chow, J.C., Watson, J.G. and Lu, Z. (1996). Descriptive Analysis of PM_{2.5} and PM₁₀ at Regionally Representative Locations During SJVAQS/AUSPEX. *Atmos. Environ.* 30: 2079-2112.
- Christine, B., Maria Cristina, F. and Satie, T. (2005). A Wintertime Study of PAHs in Fine and Coarse Aerosols in São Paulo City, Brazil. *Atmos. Environ.* 39: 3799-3811.
- Dallarosa, J.B., Mônego, J.G. and Teixeira, E.C. (2005). Polycyclic Aromatic Hydrocarbons in Atmospheric Particles in the Metropolitan Area of Porto Alegre, Brazil. *Atmos. Environ.* 39(9): 1609-1625.
- Ding, X., Wang, X.M., Xie, Z.Q., Xiang, C.H., Mai, B.X., Sun, L.G., Zheng, M., Sheng, G.Y., Fu, J.M. and PÖschl, U. (2007). Atmospheric Polycyclic Aromatic Hydrocarbons Observed Over the North Pacific Ocean and the Arctic Srea: Spatial Distribution and Source Identification. *Atmos. Environ.* 41: 2061-2072.
- Gao, C.M., Ruan, Y.Y., Feng, S.Y., Gong, Q. and Wu, L. (1993). Distribution Pattern of Polynuclear Aromatic Hydrocarbons in the Atmosphere in Huhhot City. J. Environ. Sci. 14: 79-81.
- People's Republic of China National Standard (GB3095-1996). Ambient Air Quality Standard. Admitted by Chinese Environmental Protection Bureau on 1996-01-18 and Executed on 1996-10-01.
- Gschwend, P.M. and Hites, R.A. (1981). Fluxes of Polycyclic Aromatic Hydrocarbons in Marine

and Lacustrine Sediments in the Northeastern United States. *Geochim. Cosmochim. Acta.* 45: 2359–2367.

- Guo, H.L., Lu, C.G., Yu, Q. and Chen, L.M. (2004). Pollution Characteristics of Polynuclear Aromatic Hydrocarbons on Airborne Particulate in Shanghai. *Journal of Fudan University* (*Natural Science*). 43: 1107-1112.
- Guo, H., Lee, S.C., Ho, K.F., Wang, X.M. and Zou, S.C. (2003). Particle Associated Polycyclic Aromatic Hydrocarbons in Urban Air of Hong Kong. *Atmos. Environ.* 37: 5307-5317.
- Karar, K. and Gupta, A.K. (2006). Seasonal Variations and Chemical Characterization of Ambient PM₁₀ at Residential and Industrial Sites of an Urban Region of Kolkata (Calcutta), India. *Atmos. Res.* 81(1): 36-53.
- Kulkarni, P. and Venkataraman, C. (2000). Atmospheric Polycyclic Aromatic Hydrocarbons in Mumbai, India. *Atmos. Environ.* 34: 2785-2790.
- Lee M.L., Prado G.P. and Howard J.B. (1977). Source Identification of Urban Airborne Polycyclic Aromatic Hydrocarbons by Chromatographic Mass Spectrometry and High Resolution Mass Spectrometry. *Biomed. Mass Spectrom.* 4: 182-186.
- Manoli, E., Kouras, A. and Samara, C. (2004). Profile Analysis of Ambient and Source Emitted Particle - Bound Polycyclic Aromatic Hydrocarbons from Three Sites in Northern Greece. *Chemosphere*. 56: 867-878.
- Mantis, J., Chaloulakou, A. and Samara, S. (2005). PM₁₀ Bound Polycyclic Aromatic Hydrocarbons (PAHs) in the Greater Area of Athens, Greece. *Chemosphere*. 59: 593-604.
- Mandalakis, M., Tsapakis, M., Tsoga, A. and Stefanou, E.G. (2002). Gas Particle Concentrations and Distribution of Aliphatic Hydrocarbons, PAHs, PCBs and PCDD/Fs in the Atmosphere of Athens (Greece). *Atmos. Environ.* 36: 4023-4035.
- Manoli, E., Voutsa, D. and Samara, C. (2002). Chemical Characterization and Source Identification/Apportionment of Fine and Coarse Air Particles in Thessaloniki, Greece. *Atmos. Environ.* 36: 949-961.
- Manoli, E., Kouras, A. and Samara C. (2004). Profile Analysis of Ambient and Source Emitted Particle - Bound Polycyclic Aromatic Hydrocarbons from Three Sites in Northern Greece, *Chemosphere*. 56: 867-878.
- Marr, L.C., Kirchstetter, T.W. and Harley, R.A. (1999). Characterization of Polycyclic Aromatic Hydrocarbons in Motor Vehicle Fuels and Exhaust Emissions. *Environ. Sci. Technol.* 33: 3091-3099.
- Møller, M., Alfheim, I., Larssen, S. and Mikalsen, A. (1982). Mutagenicity Ofairborne Particles in Relation to Traffic and Air Pollution Parameters. *Environ. Sci. Technol.* 16: 221-225.
- Offenberg J.H. and Baker J.E. (2000). Aerosol Size Distributions of Elemental and Organic Carbon in Urban and Over Water Atmospheres. *Atmos. Environ.* 34: 1509-1517.
- Pan, J., Zhou, C., Xu, N., Yang, Y.L. and Yin, X.C. (2003). GC/MS Analysis of Polycyclic

Aromatic Hydrocarbons in Urban Air of Qingdao City. Environ. Chem. 22: 311-312.

- Papageorgopoulou, A., Manoli, E., Touloumi, E. and Samara, C. (1999). Polycyclic Aromatic Hydrocarbons in the Ambient Air of Greek Towns in Relation to Other Atmospheric Pollutants, *Chemosphere*. 39: 2183-2199.
- Peng, L., Zeng, F.G., Chen, M.L. and Liu, Y.M. (2003). Distribution Characteristics and Source Analysis of N - Alkanes (C₁₄-C₃₁) and PAHs in Total Suspended Particulates in Urban Area of Taiyuan City. *Rock and Mineral Analysis*. 22: 206-210.
- Qi, S.H., Wang, X.M., Fu, J.M., Sheng, G.Y. and Min, Y.S. (2000). Evaluation on Pollution of Priority Polycyclic Aromatic Hydrocarbons in Aerosols at Different Function Areas of Main Cities in Pearl River Delta Economic Region. *Geochem.* 29: 337-341.
- Raiyani, C.V., Shah, S.H., Desai, N.M., Venkaiah, K., Patel, J.S., Parikh, D.J. and Kashyap, S.K. (1993). Characterization and Problems of Indoor Pollution Due to Cooking Stove Smoke. *Atmos. Environ.* 27A: 1643-1655.
- Robert, E.A., M,Brian, G.B., Josephy, P.D. and Bunce, N.J. (1999). Comparison of the Ames Salmonella Assay and Mutatox Genotoxicity Assay for Assessing the Mutagenicity of Polycyclic Aromatic Compounds in Porewater from Athabasca Oil Sands Mature Fine Tailings. *Environ. Sci. Technol.* 33: 2510-2516.
- Rogge, W.F., Hildemann, L.M., Mazurek, M.A., Cass, G.R. and Simoneit, B.R.T. (1993). Sources of Fine Organic Aerosol: 2. Non - Catalyst and Catalyst - Equipped Automobiles and Heavy Duty Diesel Trucks. *Environ. Sci. Technol.* 27: 636-651.
- Romero, R., Sienra, R. and Richter, P. (2002). Efficient Screening Method for Determination of Polycyclic Aromatic Hydrocarbons (PAHs) in Airborne Particles. Application in Real Samples of Santiago - Chile Metropolitan Urban Area. *Atmos. Environ.* 36: 2375-2381.
- Seinfeld, J.H. and Pandis, S.N. (1998). Atmospheric Chemistry and Physics: from Air Pollution to Climate Change. John Wiley & Sons Inc. New York, P. 700-703.
- Schwartz, J. (1994). Air Pollution and Daily Mortality: a Review and Meta Analysis. *Environ. Res.* 64: 36-52.
- Sharma, D.N., Sawant, A.A., Uma, R. and Cocker D.R. (2003). Preliminary Chemical Characterization of Particle - Phase Organic Compounds in New Delhi, India. *Atmos. Environ.* 37: 4317-4323.
- Sienra, M.R., Rosazza, N.G. and Préndez, M. (2005). Polycyclic Aromatic Hydrocarbons and Their Molecular Diagnostic Ratios in Urban Atmospheric Respirable Particulate Matter. *Atmos. Res.* 75: 267-281.
- Simicik, M.F., Eisenreich, S.J. and Lioy, P.J. (1999). Source Apportionment and Source/Sink Relationships of PAHs in the Coastal Atmosphere of Chicago and Lake Michigan. *Atmos. Environ.* 33: 5071-5079.
- Simonich, S.L. and Hites, R.A. (1994). Importance of Vegetation in Removing Polycyclic

Aromatic Hydrocarbons from the Atmosphere. Nat. 370 (6484): 49-51.

- Spurny, K.R. (1996). Aerosol Air Pollution Chemistry and Size Dependent Health Effects. J. Aerosol Sci. 27 (Suppl.1): 473-474.
- Takeshi, O. and Takashi, A. (2004). Spatial Distributions and Profiles of Atmospheric Polycyclic Aromatic Hydrocarbons in Two Industrial Cities in Japan. *Environ. Sci. Technol.* 38: 49-55.
- US EPA (1999). Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air Second Edition Compendium Method TO - 13A. EPA/625/R - 96/010b.
- Valavanidis, A., Fiotakis, K., Vlahogianni, T., Bakeas, E.B. and Triantafillaki, S. (2006). Characterization of Atmospheric Particulates, Particle - Bound Transition Metals and Polycyclic Aromatic Hydrocarbons of Urban Air in the Centre of Athens (Greece). *Chemosphere*. 65: 760-768.
- Venkataraman, C. and Friedlander, S.K. (1994). Source Resolution of Fine Particulate Polycyclic Aromatic Hydrocarbons Using a Receptor Model Modified for Reactivity. J. Air. Waste Manage. Assoc. 44: 1103-1108.
- Wang, J., Zhuang, M.Z., Yang, H.B., Zhang, X.M., Cao, C. and Huang Z. (2006). Pollution Characteristic of Organic Carbon and Elemental Carbon Bound to PM₁₀ in Xiamen. *Environ. Chem.* 25(4): 518-519.
- Zhou, J.B., Wang, T.G., Huang, Y.B., Mao, T. and Zhong, N.N. (2005). Size Distribution of Polycyclic Aromatic Hydrocarbons in Urban and Suburban Sites of Beijing, China. *Chemosphere*. 61: 792-799.
- Zhuang, M.Z., Yang, H.B., Wang, J., Huang, Z., Cao, C., Zhang, X.M. and Huang, B.Z. (2006). Characteristic of Inorganic Compounds and Chemical Elements in Particulate Phase. *Xiamen's Science and Technology*. 3: 42-44.

Received for review, March 19, 2007 Accepted, May 28, 2007